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UC Davis - Triglyceride Protocol

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Protocol status: Working

We use this protocol and it's working

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Keywords: Triglyceride, glycerolphosphate oxidase, triglyceride, triglyceride protocol summary, triglycerides present in the sample, peroxidase, trinder5 type color reaction, glycerol kinase, glycerol, producing hydrogen peroxide, hydrogen peroxide, dihydroxyacetone phosphate, fatty acid, free fatty acid, adenosine triphosphate, lipase, h2o2, absorbance of this dye, red colored dye, atp, phosphate, dye,

Abstract

Summary:

Triglycerides are enzymatically hydrolyzed by lipase to free fatty acids and glycerol. The glycerol is phosphorylated by adenosine triphosphate (ATP) with glycerol kinase (GK) to produce glycerol-3- phosphate and adenosine diphosphate. Glycerol-3-phosphate is oxidized by dihydroxyacetone phosphate (DAP) by glycerolphosphate oxidase producing hydrogen peroxide (H₂O₂). In a Trinder5 type color reaction catalyzed by peroxidase, the H₂O₂ reacts with 4-aminoantipyrine (4-AAP) and 3,5-dichloro-2- hydroxybenzene sulfonate (DHBS) to produce a red colored dye. The absorbance of this dye is proportional to the concentration of triglycerides present in the sample.


Materials


MATERIALS

 Calibrator **Fisher Diagnostics Catalog #TR43002**

 Reagents **Fisher Diagnostics Catalog #TR22203**

 PBS

 Microplate

 Platerreader

Reagent Preparation:

PBS – ready to use

Reagent – reconstitute with distilled water to make a 2X solution

Troubleshooting



- 1 Reconstitute powdered reagent with only 25 ml of distilled water to make a 2X solution.
- 2 Add 3 µl of calibrator and sample to each well.
- 3 Add 150 µl of PBS to each well. Read at 540 nm.

IMPORTANT: *Make sure not to add any bubbles to the wells when dispensing reagents, this will interfere with reading in the platereader.*

- 4 Add 150 µl of 2X reagent to each well. Incubate at 37°C for 5 minutes. Read at 540 nm.
- 5 Subtract blank readings from final readings. The assay will be linear so the unknown samples can be calculated as $(\text{Sample Absorbance} \div \text{Calibrator Absorbance}) \times \text{Calibrator Concentration}$.