



Version 1

UC Davis - Protein Carbonyl V.1

DOI

<https://dx.doi.org/10.17504/protocols.io.yuufwww>



Peter Havel¹

¹University of California, Davis

Mouse Metabolic Phenotyping Centers

Tech. support email: info@mmpc.org



Lili Liang

Create & collaborate more with a free account

Edit and publish protocols, collaborate in communities, share insights through comments, and track progress with run records.

Create free account



DOI: <https://dx.doi.org/10.17504/protocols.io.yuufwww>

External link: <https://mmpc.org/shared/document.aspx?id=123&docType=Protocol>

Protocol Citation: Peter Havel . UC Davis - Protein Carbonyl. [protocols.io https://dx.doi.org/10.17504/protocols.io.yuufwww](https://dx.doi.org/10.17504/protocols.io.yuufwww)

License: This is an open access protocol distributed under the terms of the [Creative Commons Attribution License](#), which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited

Created: March 05, 2019



Last Modified: May 15, 2019

Protocol Integer ID: 21108

Keywords: Protein Carbonyl, protein carbonyl colorimetric assay kit, oxidized protein, protein carbonyl summary, protein carbonyl, protein sample, convenient colorimetric assay, convenient colorimetric assay for the measurement, corresponding hydrazone, protein, protein in plasma, cayman chemical, assay, serum, dinitrophenylhydrazine, schiff base

Abstract

Summary:

Cayman Chemical's Protein Carbonyl Colorimetric Assay Kit is a convenient colorimetric assay for the measurement of oxidized proteins. Protein samples are derivatized by making use of the reaction between 2,4-dinitrophenylhydrazine (DNPH) and protein carbonyls. Formation of a Schiff base produces the corresponding hydrazone which can be analyzed spectrophotometrically at 360-385 nm. This assay can be used to measure oxidized protein in plasma, serum, cell lysates, and tissue homogenates.

Materials

MATERIALS

⊗ Assay Kit **Cayman Chemical Company Catalog #10005020**

⊗ HCl

⊗ DNPH

⊗ TCA

⊗ EtOH

⊗ Ethyl Acetate

Note:

Cayman Chemical **RRID:SCR_008945**



- 1 Transfer 200 μ l of sample to two 2 ml plastic tubes. One tube will be the sample tube (S#) and the other will be the control tube (C#).
- 2 Add 800 μ l of DNPH to the sample tube and add 800 μ l of 2.5 M HCl to the control tube.
- 3 Incubate both tubes (S# & C#) in the dark at room temperature for one hour. Vortex each tube briefly every 15 minutes during the incubation.
- 4 Add 1 ml of 20% TCA to each tube and vortex. Place tubes on ice and incubate for five minutes.
- 5 Centrifuge tubes at 10,000 xg for 10 minutes at 4°C in a microcentrifuge.
- 6 Discard the supernatant and resuspend the pellet in 1 ml of 10% TCA. Place tubes on ice and let sit for five minutes.
- 7 Centrifuge tubes at 10,000 xg for 10 minutes at 4°C in a microcentrifuge.
- 8 Discard the supernatant and resuspend the pellet in 1 ml of (1:1) Ethanol/Ethyl Acetate mixture. Manually suspend pellet with spatula, vortex thoroughly, and centrifuge tubes at 10,000 xg for 10 minutes at 4°C in a microcentrifuge.
- 9 Repeat Step 8 two more times.
- 10 After the final wash, resuspend the protein pellets in 500 μ l of guanidine hydrochloride by vortexing.
- 11 Centrifuge tubes at 10,000 xg for 10 minutes at 4°C in a microcentrifuge to remove any left over debris.
- 12 Transfer 220 μ l of supernatant from the sample (S#) tube to two wells of the 96-well plate.
- 13 Transfer 220 μ l of supernatant from the control (C#) tube to two wells of the 96-well plate.
- 14 Measure the absorbance at a wavelength between 360-385 nm using a plate reader.

15 Calculation

1. Calculate the average absorbance of each sample and control.
2. Subtract the average absorbance of the controls from the average absorbance of the samples. This is the Corrected Absorbance (CA).
3. Determine the concentration of the carbonyls by inserting the corrected absorbance into the following equation:

$$\text{Protein Carbonyl (nmol/ml)} = [(CA)/(*0.011 \mu\text{M}^{-1})](500 \mu\text{l}/200 \mu\text{l})$$

*The actual extinction coefficient for dinitrophenylhydrazine at 370 nm is $22,000 \text{ M}^{-1}\text{cm}^{-1}$ ($0.022 \mu\text{M}^{-1}\text{cm}^{-1}$). This value has been adjusted for the pathlength of the solution in the well.