Scanning electron microscopy (SEM) protocol for imaging living materials

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ABSTRACT
Scanning electron microscopy (SEM) can be used to image cells and colonies immobilized inside hydrogels after supercritical carbon dioxide (CO2) extraction. Supercritical CO2 extraction can also be used on suspension cells after filtering the sample onto a 0.2 μM filter attached into the extractors carriers. This protocol gives an overview on how different techniques can be used to characterize triblock copolymer hydrogels.

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KEYWORDS
SEM, living materials, hydrogels, triblock copolymers, immobilized microbial cells, yeast, polymer degradation, supercritical carbon dioxide extraction, scanning electron microscopy

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CREATED
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**Materials Text**

**Reagents:**
- formaldehyde
- Ethanol, 99.5 - 100 %
- MilliQ water, sterile
- 0.2 M phosphate buffer (20.44 g of Na₂HPO₄ and 6.72 g of NaH₂PO₄ per litre)
- liquid N₂

**Supplies:**
- falcon tubes/ependorfs/glass vials
- scalpels
- 12.5 mm aluminum SEM pin stubs
- conductive double sided carbon tabs/tape
- sharpie marker

**Safety Warnings**
Formaldehyde (FA) is toxic and should handled accordingly. Wear protective gear!

N₂ cooled scalpels can break during sample cutting. Wear protective eyewear!

Supercritical CO₂ extraction involves high pressure. Do not leave extractor unattended while chamber temperature is rising.

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**Sample Fixation**

1. Prepare fixation solution (3.7 % volume formaldehyde in 0.1 Molarity (M) phosphate buffer) for 30m

2. Submerge sample into the fixation solution and incubate at Room temperature for 24:00:00

3. Replace fixation solution and incubate at Room temperature for 24:00:00

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**Sample Dehydration**

4. Prepare ethanol (EtOH) dilutions in milli-Q water as indicated below for 30m

5. Samples are dehydrated at Room temperature in an ascending EtOH series (40 – 90 %, 10 % steps; 96 %, 99.5 %).

Submerge sample into EtOH solution, let it incubate (minimum 02:00:00 per step), change the EtOH solution.

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**EtOH steps:**
1. 40% volume EtOH
2. 50% volume EtOH
3. 60% volume EtOH
4. 70% volume EtOH
5. 80% volume EtOH
6. 90% volume EtOH
7. 96% volume EtOH
8. 99.5% volume EtOH (Overnight)
9. 99.5% volume (for storage)

**Supercritical CO₂ extraction**

6. Cool the critical point dryer (E3100, Quorum Technologies) to 15°C with a thermostat (Proline RP 1845, LAUDA) using thermostat external temperature probe.

7. Connect the critical point dryer outlet to a bottle containing ETOH (half full) under fume hood (it is used to capture residues during extraction and to estimate the gas release speed).

8. Open the critical point dryer and mount the samples. Close the critical point dryer according to producers instructions.

9. Open CO₂ inlet and fill the chamber with liquid CO₂.

10. Slightly open the outlet and purge the chamber for 00:05:00 (bubbling inside the external ETOH bottle should not be too intensive).
    After purging close the outlet first then the inlet (to avoid pressure drop inside the chamber).

11. The chamber should be purged with fresh CO₂ 6 – 8 times in 30 – 60 min intervals (let the ETOH diffuse out of the structure and purge it out of the chamber, use shorter intervals at the beginning of this process).
    - open the CO₂ inlet, then slightly open the outlet
    - purge for 00:05:00
    - close the outlet, then close the CO₂ inlet
    - repeat 6-8 times in 30 – 60 min intervals

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Increase the thermostat temperature to **37 °C** (inlet and outlet of the chamber should be closed at this point).

Do not leave the critical point dryer unattended while the temperature is rising as the pressure can exceed the safety limit of the chamber.

Control the internal pressure so it does not exceed 110 bar by opening the chamber outlet (should be done slowly as too fast gas release can cool the reactor and turn supercritical state back to liquid state).

Adjust the outlet so that the pressure gauge stays stable around 105 bar as the temperature is rising

Leave the outlet open as it is, when **37 °C** is achieved (do not open the outlet more, as the faster gas release can cool the reactor).

Step 12 includes a Step case.

### SLOW

![Image A](image1.png) ![Image B](image2.png)

**Figure 1** Slow pressure release (A) vs. fast pressure release (B)

Leave the outlet open until the chamber is ready to be opened (**Overnight**).

As the pressure drops so does the bubbling.

Adjust the outlet so that there is always slight bubbling (do not over do it as it can result in pore formation **figure 1B**)

Before opening the chamber remove outlet tube from ETOH bottle that is situated under the hood (to avoid sucking ETOH into the chamber while opening it).

Remove samples from the chamber and store in a sealable container (ependorf, glass vial, falcon tube)

**Sample cutting and mounting**

Attach conductive double sided carbon tabs/tape on aluminum SEM pin stubs and then lable them with a sharpie marker.
Figure 3: Sample cutting - exposing cell-material interactions in hydrogels. Varying the sample and the blade temperature together with the speed of cutting can be used to demonstrate various aspects of LMs. A combination of sample and scalpel cooling (~20 s) together with fast incisions results in the most accurate SEM images in terms of polymeric material and colony localization (A, B), but with this technique it is impossible to evaluate the colony size and shape because of the unknown location of the obtained cross-section in respect to the colony. A shorter duration of sample and scalpel cooling (~10 s) together with slow incision highlights biologically relevant information such as cell-polymer encapsulations (Figure 5 A - C) and colony size and shape (C, D) but results in cutting marks across the polymer (D). Different sample cuttings and resulting images: samples prepared with longer cooling of sample and scalpel showing relatively smooth cuts (A, B). Samples prepared with short sample and scalpel cooling showing clear colonies (C, D).

16.1 Fast incision (Figure 3: A, B) - for acquiring artifact-free cross-sections

Immerse the sample with forceps and scalpel into liquid N\textsubscript{2} for 00:00:20 and instantly cut with fast incision (N\textsubscript{2} cooled scalpels can break during sample cutting. Wear protective eyewear!).

16.2 Slow incision (Figure 3: C, D) - for acquiring information of colony-material interactions, colony size and shape

Immerse the sample with forceps and scalpel in liquid N\textsubscript{2} for 00:00:10 and cut after 00:00:03 at room temperature with slow incision.

17 Using forceps, pick up the cut sample and gently press it onto the two-sided carbon tape.

18 Coat the sample with a \textbf{7.5 nm} gold layer using a high vacuum sputter coater (EM ACE600, Leica Microsystems).

Gold-coated samples were imaged with a tabletop scanning electron microscope (TM3000, Hitachi).
The imaging was done under a high vacuum and 15 kV accelerating voltage.

Figure 2. Fast CO2 release at the end of supercritical CO2 extraction process generates local stresses in the gel that lead to mechanical changes resulting in a foam-like structure (A) and the separation of regions in the material (B, C) based on the local structural and mechanical properties. The amount and size of the generated pores depends on the speed of gas release. This method is a valuable tool for studying gas retention in the material, the thickness of thin film coatings around colonies (D) and possible degradation of the matrix (E, F). F127-BUM samples with (F) and without (E) protease inhibitor showing that when inhibitor was missing, the gas did not get trapped with fast CO2 release.

13 Let the chamber pressure drop until it reaches 60 bar (at this point CO2 has turned from supercritical state into gas state and opening the outlet will not result in temperature drop).

14 Remove the outlet from the external EtOH bottle as it can splatter under the hood at this step.

Open outlet so the remaining pressure is released within 3-5 min (the generated pores depends on the speed of gas release)

15 Before opening the chamber remove outlet tube from EtOH bottle that is situated under the hood (to avoid sucking EtOH into the chamber while opening it).

Remove samples from the chamber and store in a sealable container (ependorf, glass vial)
16. Attach conductive double sided carbon tabs/tape on aluminum SEM pin stubs and then label them with a sharpie marker.

17. Figure 3: Sample cutting - exposing cell-material interactions in hydrogels. Varying the sample and the blade temperature together with the speed of cutting can be used to demonstrate various aspects of LMs. A combination of sample and scalpel cooling (~20 s) together with fast incisions results in the most accurate SEM images in terms of polymeric material and colony localization (A, B), but with this technique it is impossible to evaluate the colony size and shape because of the unknown location of the obtained cross-section in respect to the colony. A shorter duration of sample and scalpel cooling (~10 s) together with slow incision highlights biologically relevant information such as cell-polymer encapsulations (Figure 5 A - C) and colony size and shape (C, D) but results in cutting marks across the polymer (D). Different sample cuttings and resulting images: samples prepared with longer cooling of sample and scalpel showing relatively smooth cuts (A, B). Samples prepared with short sample and scalpel cooling showing clear colonies (C, D).

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Immerse the sample with forceps and scalpel in liquid N\(_2\) for 00:00:10 and cut after 00:00:03 at room temperature with slow incision.

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Sputter Coating 1h

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19 Coat the sample with a 7.5 nm gold layer using a high vacuum sputter coater (EM ACE600, Leica Microsystems).

20 Gold-coated samples were imaged with a tabletop scanning electron microscope (TM3000, Hitachi). The imaging was done under a high vacuum and 15 kV accelerating voltage.