

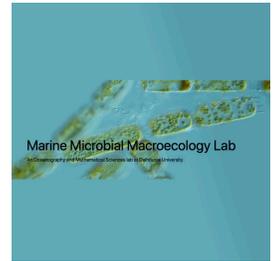
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## Measurement of dissolved carbohydrate V.2

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**Protocol status:** Working

**We use this protocol and it's working**

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**Keywords:** Dissolved total carbohydrate, Dissolved polysaccharide, Dissolved monosaccharide, TPTZ method, Ferricyanide, hydrolysis, dissolved carbohydrate measurement, dissolved carbohydrate sample, dissolved carbohydrate, dissolved monosaccharide measurement, carbohydrate measurement, carbohydrate sample, dissolved monosaccharide, dissolved polysaccharide, monosaccharide measurement, final h2so4 molarity, alkalized hydrolysate, ferricyanide solution, hydrolysate, absorbance of tptz, amino sugar, absorbance, monosaccharide

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## Abstract

Here we describe a protocol to measure the dissolved carbohydrate, including total dissolved monosaccharides and total dissolved polysaccharides.

For total dissolved carbohydrate measurement, freeze-dried dissolved carbohydrate samples are initially vortexed in 9 M H<sub>2</sub>SO<sub>4</sub> for 15 s. The solution is diluted for a final H<sub>2</sub>SO<sub>4</sub> molarity of 1.6 M and hydrolyzed for 3 hours at 90 °C. The hydrolysate is alkalized by adding 12 M NaOH to the hydrolysate, the ratio of [H<sup>+</sup>] from hydrolysate to [OH<sup>-</sup>] from NaOH is 0.82. The alkalized hydrolysate is oxidized by ferricyanide solution. The absorbance of TPTZ-Fe<sup>2+</sup> complex is measured in microtiter plate at 595 nm.

For total dissolved monosaccharide measurement, freeze-dried dissolved carbohydrate samples are alkalized by 12 M NaOH and then oxidized by ferricyanide solution. The absorbance of TPTZ-Fe<sup>2+</sup> complex is measured in microtiter plate at 595 nm.

Our method has shown high reproducibility in aldohexoses, ketohexoses, deoxysugars, aldopentoses, uronic acid and amino sugars. The low limit of detection is 5 ng C/mL.

## Protocol materials

⊗ D-glucose Merck MilliporeSigma (Sigma-Aldrich) Catalog #G8270-100G

⊗ NaOH Fisher Scientific Catalog #BP359-500

⊗ Na<sub>2</sub>CO<sub>3</sub> VWR International (Avantor) Catalog #97061-972

⊗ K<sub>3</sub>[Fe(CN)<sub>6</sub>] Fisher Scientific Catalog #AC424120050

⊗ TPTZ Merck MilliporeSigma (Sigma-Aldrich) Catalog #T253-5G

⊗ Sodium acetate anhydrous Fisher Scientific Catalog #BP333-500

⊗ Citric acid Merck MilliporeSigma (Sigma-Aldrich) Catalog # 251275-500G

⊗ Acetic acid Fisher Scientific Catalog #M1000632500

## Troubleshooting

## Safety warnings

- ! Ferric waste should be disposed into trace metal waste container.  
Waste acid should be neutralized before disposed into sink.

## Before start

- Combust vials, glass centrifuge tubes, glass filter funnel and filter flask at  500 °C  06:00:00
- Combust glassware to hold 18 M H<sub>2</sub>SO<sub>4</sub> at  500 °C  06:00:00
- Combust GF/F filter at  450 °C  04:00:00
- Combust filter holder at  500 °C  02:00:00
- All vial and tube caps are acid-washed and dried.

## Sample collection

12h

1

### Note

For lab-cultivated samples, ensure to collect dissolved carbohydrate samples starting from Day 1 as the initial dissolved carbohydrate value.

2

Precombust clear vials at  500 °C for  06:00:00

6h

### Equipment

Glass Vials PTFE/SILiCone SEPTA Clear	NAME
16 mL	TYPE
Thermo Scientific	BRAND
B7990-4	SKU

Acid wash caps:

### Equipment

Screw Vial Convenience Kit, 12mL solid top PTFE cap	NAME
Thermo Scientific	BRAND
B7800-12A	SKU

3

Filter microalgae sample and collect the filtrate, using gentle vacuum pressure (130 mm Hg).



4 Transfer 5 mL filtrate into 12 mL clear vial and store at -20 °C .

#### Note

Three vials for total dissolved monosaccharide and another three vials for total dissolved carbohydrate measurement.  
Volume of filtrate must be precisely measured.

5 Transfer 5 mL blank media into 12 mL clear vial, in duplicate, and store at -20 °C

#### Note

Blank is essential especially when filtered seawater is used as media base.

6 Freeze-dry samples before measurement.

## Glucose standards

7 Primary standard solution

7.1 In a 2 mL microtube, weigh 1 ~ 2 mg D-glucose

D-glucose Merck MilliporeSigma (Sigma-Aldrich) Catalog #G8270-100G

7.2 Add Milli-Q for a final concentration of 1 mg/mL (>600 µL).

8 Secondary standard for total dissolved carbohydrate

8.1 Transfer 45 µL primary solution into a 2 mL microtube

8.2 Add 955 µL Milli-Q and then vortex well

8.3 In 10 mL centrifuge tubes, prepare the following standard solutions:

SD	Secondary solution (uL)	Milli-Q (uL)
TCHO-SD1	0	100
TCHO-SD2	20	80
TCHO-SD3	40	60
TCHO-SD4	60	40
TCHO-SD5	80	20
TCHO-SD6	100	0

9 Secondary standard for total dissolved monosaccharide

9.1 Add  10 µL primary solution into a 2 mL microtube

9.2 Add  990 µL Milli-Q and then vortex for a good mix

9.3 In 12 mL amber vials, prepare the following standard solutions:

SD	Secondary solution (uL)	Milli-Q (uL)	12 M NaOH (uL)
MCHO-SD1	0	984	16
MCHO-SD2	10	974	16
MCHO-SD3	20	964	16
MCHO-SD4	50	934	16
MCHO-SD5	100	884	16
MCHO-SD6	150	834	16



## Equipment

### Storage Vials and Closures

12 mL amber

Thermo Scientific

B7800-12A

VWR 66030-686

NAME

TYPE

BRAND

SKU

SPECIFICATIONS

## Hydrolysis of total dissolved carbohydrate

20s

10 Prepare water bath 95 °C

11 Add 100 µL Milli-Q to each tube with freeze-dried sample.

12 Use the sonicator to fully suspend the pellets in the water, ensuring minimal particles remain on the sides of the vial.

13 Use reverse pipetting, add 100 µL 18 M H<sub>2</sub>SO<sub>4</sub> to standard solution/sample, immediately vortex for 00:00:15 (monitored by timer or stopwatch)

15s

### Note

Do not cap the centrifuge tube!

14 Add 900 µL Milli-Q, tightly cap the centrifuge tube, and vortex for 00:00:05 .

5s

15 Place tube into water bath, log the time.

Note

Three-hour hydrolysis duration for each sample/blank/standard should be accurately monitored.

16 After all samples are in the water bath, reduce temperature to  $90\text{ }^{\circ}\text{C}$  .

17 Label amber vials for TPTZ measurement with white oil based sharpie.

# of vials = # of samples + # of blanks + # of standards

18 As soon as hydrolysis duration reaches 3 hours, remove the tube from water bath, cool in the tap water bath with ice to quickly stop hydrolysis.

19 Keep all hydrolysate (including standards and samples) in a dark cabinet at  $\text{Room temperature}$  .

## Prepare TPTZ reagents

20 12 M NaOH

20.1 Add 15 mL Milli-Q water into a 50 mL Falcon tube.

20.2 Add  $12\text{ g}$  NaOH pellet into the water, swirl and have the pellets completely dissolved, let it cool down to  $\text{Room temperature}$  .

20.3 Transfer the solution into a 25 mL PP volumetric flask, rinse the tube three times by small amount of Milli-Q and combine the rinsed water into flask, top with Milli-Q water to 25 mL.

21 Alkaline solution for potassium ferricyanide

Dissolve  $400\text{ mg}$  NaOH and  $20\text{ g}$   $\text{Na}_2\text{CO}_3$  in volumetric flask and top to 1 L by Milli-Q. Store at room temperature.

 NaOH Fisher Scientific Catalog #BP359-500

  $\text{Na}_2\text{CO}_3$  VWR International (Avantor) Catalog #97061-972

22 Sodium acetate solution

22.1 Dissolve 164 g sodium acetate, 42 g citric acid and 300 g acetic acid in a 1 L volumetric flask and top to 1 L with Mill-Q water.

Note

1. In this solution, sodium acetate, citric acid and acetic acid is 2 M, 0.2 M and 5 M respectively.
2. Add sodium acetate into the dry volumetric flask first. Sodium acetate is highly hygroscopic, the absorbance of moist hardens the powder into a bulk and clogs the neck of flask.

⊗ Sodium acetate anhydrous Fisher Scientific Catalog #BP333-500

⊗ Citric acid Merck MilliporeSigma (Sigma-Aldrich) Catalog # 251275-500G

⊗ Acetic acid Fisher Scientific Catalog #M1000632500

22.2 Store at room temperature.

22.3 Dispense solution by serological pipet to avoid having salt precipitated around sealing surface of the bottle.

23 3 M acetic acid

Weigh 180 g acetic acid in fumehood, transfer the acid into volumetric flask, top to 1 L with Milli-Q water. Store at room temperature.

## TPTZ method

40m

24 Prepare boiling bath

25 TPTZ reagents

25.1 Potassium ferricyanide (Reagent A)

Weigh 23 mg potassium ferricyanide and transfer into a 100 mL amber reagent bottle. Add 100 mL alkaline solution, vortex until powder is completely dissolved. It is stable for two weeks at room temperature.

⊗  $K_3[Fe(CN)_6]$  Fisher Scientific Catalog #AC424120050

### Equipment

<b>Reagent bottle</b>	NAME
100 mL, amber	TYPE
VWR	BRAND
14216-240	SKU

#### 25.2 Ferric chloride (Reagent B)

Ferric chloride hexahydrate is in spherical shape. It is hard to weigh exact 54 mg for a 100 mL solution. Pick a very small ferric chloride ball and log the weight. Transfer the ball into a 100 mL amber reagent bottle. Calculate the acetate solution required. Add acetate solution into the amber bottle, vortex until the ball is completely dissolved.

$$V_{\text{acetate}} = 100 \times W_{\text{actual}} / 54$$

#### Note

This reagent needs to be prepared right prior to analysis. It can only be stable for no more than two days.

#### 25.3 TPTZ (Reagent C)

Estimate the total volume required for the assay: 2 mL X (standard # + blank # + sample #)

For each 100 mL TPTZ reagent, weigh and transfer 78 mg TPTZ into an amber reagent bottle, add 100 mL acetic acid solution, vortex until the powder is completely dissolved.

 TPTZ Merck MilliporeSigma (Sigma-Aldrich) Catalog #T253-5G

#### Note

This solution is stored at room temperature and stable for one week.



26 Total dissolved carbohydrate samples

26.1 Use reverse pipetting, transfer  750  $\mu\text{L}$  hydrolysate of standard/sample to amber vial.

26.2 Reverse pipetting, add  250  $\mu\text{L}$  12 M NaOH and vortex.

27 Total dissolved monosaccharide samples

27.1 Add  1200  $\mu\text{L}$  Milli-Q into the tube with freeze-dried sample

27.2 Use reverse pipetting, transfer  984  $\mu\text{L}$  solution to amber vial.

27.3 Add  16  $\mu\text{L}$  12 M NaOH and vortex.

28 In a room with dim light, add  1 mL Reagent A into each amber vial.

29 Tightly cap the vial and vortex.

30 Keep in a boiling water bath for  00:10:00

10m

31 Remove boiling bath from the heat, keep all vials in the hot water and move them into the room with dim light.

32 Add  1 mL Reagent B and  2 mL Reagent C into the vial and vortex.

33 Shake at  Room temperature for  00:30:00 .

30m

34 Under dim light, using reverse pipetting, load 250 uL of blanks, standards, and samples into the microplate (duplicate).

Load column by column. After one column has been loaded, immediately cover the column with a lid, which has a black membrane on the top to protect sample from light.

		1	2	3	4	5	6	7	8	9	10	11	12
	A	MCHO-SD1	MCHO-SD1										
	B	MCHO-SD2	MCHO-SD2										
	C	MCHO-SD3	MCHO-SD3										
	D	MCHO-SD4	MCHO-SD4										
	E	MCHO-SD5	MCHO-SD5										
	F	MCHO-SD6	MCHO-SD6										
	G												
	H												

Microplate layout for dissolved monosaccharide samples

		1	2	3	4	5	6	7	8	9	10	11	12
	A	TCHO-SD1	TCHO-SD1										
	B	TCHO-SD2	TCHO-SD2										
	C	TCHO-SD3	TCHO-SD3										
	D	TCHO-SD4	TCHO-SD4										
	E	TCHO-SD5	TCHO-SD5										
	F	TCHO-SD6	TCHO-SD6										
	G												

	1	2	3	4	5	6	7	8	9	10	11	12
H												

Microplate layout for total dissolved carbohydrate samples

35 Read in microplate reader:

Shake for 5 s at 600 rpm in a continuous and high force mode

Read endpoint 595 nm with a measurement time 100 ms

### UV/VIS spectra (optional)

36 Hydrolysate of the samples

36.1 Load  200 µL hydrolysate into microplate.

36.2 Blank:  
TCHO-SD1

37 Monosaccharide solutions

37.1 Load  200 µL solution into microplate.

37.2 Blank:  
Milli-Q

38 Scan UV/VIS spectra from 200 to 400 nm at a step of 1 nm.

### Calculation

39 Total dissolved carbohydrate

39.1 Subtract the average absorbance of blank (0 ug glucose) from the absorbance of each standard for total dissolved carbohydrate.

39.2 Obtain standard curve by plotting blank subtracted absorbance ( $Abs'$ ) versus carbon ( $\mu M C$ )

$$Abs' = a * C_{(\mu M)} + b$$

39.3 Subtract the average absorbance of blank (0 ug glucose) from the absorbance of each sample

39.4

$$C_{(\mu M)} = (Abs' - b)/a$$

$$TCHO_{(\mu MC)} = C * (1.1/5)/0.75$$

40 Total dissolved monosaccharide

40.1 Subtract the average absorbance of blank (0 ug glucose) from the absorbance of each standard for total dissolved monosaccharide.

40.2 Obtain standard curve by plotting blank subtracted absorbance ( $Abs'$ ) versus carbon ( $\mu M C$ )

$$Abs' = a * C_{(\mu M)} + b$$

40.3 Subtract the average absorbance of blank (0 ug glucose) from the absorbance of each sample

40.4

$$C_{(uM)} = (Abs' - b)/a$$

$$MCHO_{(uMC)} = C * (1.2/5)/0.984$$

41 Total dissolved polysaccharide

$$PCHO_{(uMC)} = TCHO_{(uMC)} - MCHO_{(uMC)}$$

## Waste disposal

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42 All hydrolysate and TPTZ reagent C need to be neutralized by soda before disposed into the sink.

43 TPTZ reagent B is collected in trace metal waste container.