

Apr 06, 2023

Version 2

Estimate phospholipids from microalgae V.2

DOI

dx.doi.org/10.17504/protocols.io.q26g74r19gwz/v2



Ying-Yu Hu¹, Zoe V. Finkel¹

¹Dalhousie University

Marine Microbial Macroecology Lab Tech. support email: ruby.hu@dal.ca



Ying-Yu Hu

Dalhousie University

Create & collaborate more with a free account

Edit and publish protocols, collaborate in communities, share insights through comments, and track progress with run records.

Create free account





DOI: https://dx.doi.org/10.17504/protocols.io.q26g74r19gwz/v2

Protocol Citation: Ying-Yu Hu, Zoe V. Finkel 2023. Estimate phospholipids from microalgae. **protocols.io** https://dx.doi.org/10.17504/protocols.io.q26g74r19gwz/v2 Version created by Ying-Yu Hu



License: This is an open access protocol distributed under the terms of the **Creative Commons Attribution License**, which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited

Protocol status: Working

We use this protocol and it's working

Created: February 16, 2023

Last Modified: April 06, 2023

Protocol Integer ID: 77080

Keywords: phospholipids, high temperature dry combustion, estimate phospholipids from microalgae, phospholipid, complete conversion of phospholipid, recovery rate of phospholipid, total lipids from microalgae, phospholipid, remaining lipid extract, lipid extract, use of glass vial, lipid, traditional dry combustion, using glassware, total lipid, resulting orthophosphate, capped glass vial, glass vial, acid digestion method, microalgae

Funders Acknowledgements:

Simons Foundation Grant ID: 549937 Simons Foundation Grant ID: 723789



Abstract

Here we describe a protocol to estimate phospholipids from microalgae.

After extracting and measuring the total lipids from microalgae, the remaining lipid extract is dried using a nitrogen flow, followed by drying with magnesium sulfate at 90°C. However, it has been observed that traditional dry combustion at 500°C only decomposes approximately 50% of phospholipids (Hu et al., 2022). To achieve complete conversion of phospholipids to pyrophosphate, a temperature of around 800°C is required, but such high temperatures cannot be used with glassware. As the acid digestion method involves using only 500 µL of 0.2 M HCl, which must be placed in tightly capped glass vials to prevent concentration changes due to evaporation, combustion must be carried out using glassware instead of crucibles. It should be noted that the recovery rate of phospholipids is around 80% when combusted at 650°C, but this recovery rate is consistent, making the use of glass vials applicable. Therefore, we recommend using 650°C to combust phospholipids and using 80% to correct the final results.

The resulting ash is digested using 0.5 mL of 0.2 M HCl for 30 minutes at 90°C. After digestion, the resulting orthophosphate is detected by mixing the sample with a combination of molybdate and ascorbic acid to produce molybdenum blue, as described in Chen's work (1956).

Citation

P.S. Chen, T.Y. Toribara and Huber Warner. Microdetermination of Phosphorus. Anal. Chem..

https://doi.org/10.1021/ac60119a033

LINK

Citation

Ying-Yu Hu, Andrew J. Irwin, Zoe V. Finkel (2022). Improving quantification of particulate phosphorus. Limnology and Oceanography: Methods.

https://doi.org/10.1002/lom3.10517

LINK



Protocol materials

- Magnesium sulfate anhydrous Fisher Scientific Catalog #M65500
- X 12 N Hydrochloric acid
- 🔀 18M sulfuric acid
- Ammonium molybdate Merck MilliporeSigma (Sigma-Aldrich) Catalog #09878-100G
- Ascorbic acid Merck MilliporeSigma (Sigma-Aldrich) Catalog #A5960-100G
- Potassium dihydrogen orthophosphate ACP Chemicals Catalog #P-4550

Troubleshooting



Prepare phospholipids sample

Dry remaining organic phase extract of total lipids at 37 °C under a stream of N₂ gas (<2 psi)

Phosphate primary standard

2h

- 2 KH_2PO_4 primary standard stock solution ($\approx 1 \text{ mM}$)
- Potassium dihydrogen orthophosphate ACP Chemicals Catalog #P-4550
- 2.1 Transfer about 1 g KH₂PO₄ into a beaker, cover the beaker with foil
- 2.2 Place the beaker into an oven, dry KH₂PO₄ at \$\mathbb{8}\$ 110 °C for at least \$\infty\$ 02:00:00

2h

- 2.3 Move KH₂PO₄ into a vacuum desiccator, allow KH₂PO₄ to cool to room temperature
- 2.4 Dissolve around $\perp 0.136 \text{ g}$ dried KH₂PO₄ in $\perp 1 \text{ L}$ MilliQ water.
 - Use 1 L volumetric flask
 - Take notes of the actual weight of KH₂PO₄ for final concentration of standard stock solution
- 2.5 Transfer standard stock solution into a 1 L bottle and store in the fridge.

Note

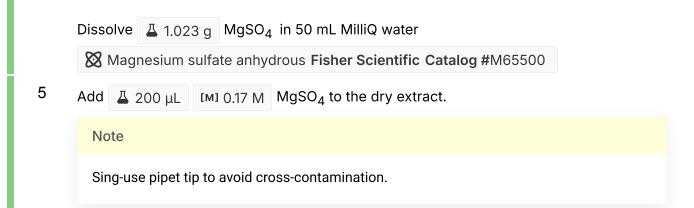
This stock solution lasts quite a long time, unless there is evidence for growth of algae or other extraneous biotic material.

High temperature dry combustion

9h

- 3 Use diamond pen to engrave the sample vials with numbers. Log number and sample code.
- 4 [M] 0.17 M MgSO₄ reagent:





6 Cover the uncapped vials with foil and place in the oven at \$\mathbb{\ until samples are completely dry.



Note

Remove samples out of the oven as soon as they are dried. If muffle furnace is not available, keep samples in vacuum desiccator.

7 Combust dried samples at \$\\$650 \circ\$ for \$\\$09:00:00

9h



Equipment	
Muffle furnace	NAME
F30428C	TYPE
Thermo	BRAND
10-505-13	SKU

Note

8 Allow samples to gradually cool down in the muffle furnace.

Digestion

9 [M] 0.2 M HCl reagent:

In a reagent bottle, dissolve one part of [M] 12 N HCl in 59 parts of MilliQ water

🔀 12 N Hydrochloric acid

Note

Volume of HCl_0.2M_mL = (0.5_mL) X (#Sample + #Blank)

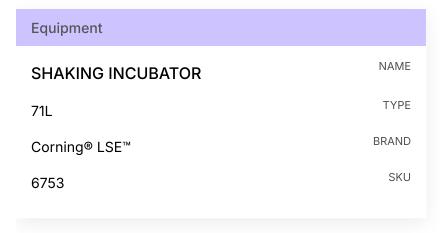
10



- 11 Add 4 0.5 mL [M] 0.2 M HCl to each vial.
- 12 Tightly cap the vial and vortex.
- 13 Place vials in the oven for 00:30:00
- 14 Cool samples down to | Room temperature

Preparing standard working solutions

15 Preheat shaker/incubator to 37 °C



- 16 Standard working solutions and reagents can be prepared during sample digestion.
- 17 Standard working solution

	Standard	Primary (uL)	MilliQ (uL)
	S1	0	1000
	S2	5	995

30m



Standard	Primary (uL)	MilliQ (uL)
S3	10	990
S4	20	980
S5	50	950
S6	100	900
S7	150	850
S8	200	800

18 Transfer Δ 500 μ L of each standard working solution to 2 mL microtube.

Preparing working reagents

All reagents are freshly prepared before colorimetric measurement.

20 [M] 2.5 % ammonium molybdate reagent:

Weigh $\begin{tabular}{ll} $\underline{\begin{tabular}{ll} $\underline{\begin{tabular} $\underline{\begin{tabular}{ll} $\underline{\begin{tabular}{ll} $\underline{\begin{tabular} \underline

Cap and shake until totally dissolved.

Ammonium molybdate Merck MilliporeSigma (Sigma-Aldrich) Catalog #09878-100G

21 [м] 10 % ascorbic acid reagent (avoid light exposure):

🔀 Ascorbic acid Merck MilliporeSigma (Sigma-Aldrich) Catalog #A5960-100G

22 [M] 6 N (3 M) sulfuric acid reagent:



Carefully add 1 part M concentrated sulfuric acid into 5 part MilliQ water

X 18M sulfuric acid

- 23 Calculate the volume of molybdate-ascorbic reagent: Total volume of reagent_mL = (0.5 mL) X (#standard working solution + #samples + #blanks)
- 24 Mix the reagents into Falcon tube:

Reagent	Parts as in volume
MilliQ	2
6N sulphuric acid	1
2.5% ammonium molybdate	1
10% ascorbic acid	1

Colorimetric measurement

3h

25 Add 4 500 µL reagent to each standard, sample (in the vial) and blank, starting from blanks, including blank for standards and blank for samples.

Equipment	
Finntip Stepper Tips	NAME
5 mL	TYPE
Thermo Scientific	BRAND
9404200	SKU

Note

Before dispensing the reagent, wipe or dab the liquid drop on the outside of the tip, avoid wiping the open tip.

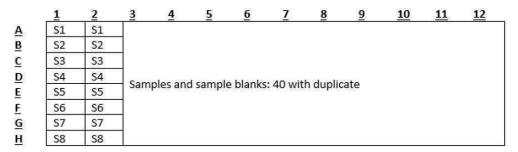


26 Vortex.

27 Incubate at 37 °C for 3:00:00 while shaking at 150 rpm

3h

28 Load microplate with 250 ul reactant from each tube, duplicate.



Example of loading the microplate

29 Read plate in microplate reader

А	В
Shake duration	00:00:05
Shaking type	Continuous
Shaking force	High
Shaking speed [rpm]	600
Wavelength [nm]	820
Use transmittance	No
Pathlength correction	No
Measurement Time [ms]	100



Equipment

Varioskan LUX Multimode Microplate Reader NAME

BRAND Thermo Fisher

SKU VL0L00D0

Calculation

3h

- 30 Subtract the average absorbance at 820 nm of the blank standard replicates from the absorbance at 820 nm of all other standard working solutions.
- 31 Subtract the average absorbance at 820 nm of the blank sample (i.e. blank filter) replicates from the absorbance at 820 nm of all other individual samples.
- 32 Prepare a standard curve by plotting the average blank-corrected 820 nm absorbance for each standard working solution versus its concentration in uM. Molar Mass of KH2PO4: 136.086 g/mol
- 33 Use the standard curve to determine the orthophosphate concentration of each unknown sample by using its blank-corrected 820 nm absorbance.
- 34 $(P_{measured})_{umol/sample} = (orthophosphate)_{um} X (V_{HCl})_{mL} X (0.001)$ $(P_{corrected})_{umol/sample} = (P_{measured}) / 0.8$ Where, 0.8 is the average recovery of phospholipids after a high temperature dry combustion at 4 650 °C .
- 35 (Phospholipids)_ug/sample = $(P_{corrected})X30.97/(0.01X4.3)$



Citations

P.S. Chen, T.Y. Toribara and Huber Warner. Microdetermination of Phosphorus https://doi.org/10.1021/ac60119a033

Ying-Yu Hu, Andrew J. Irwin, Zoe V. Finkel. Improving quantification of particulate phosphorus https://doi.org/10.1002/lom3.10517