OCT 04, 2019

Determining Chlorophyll Concentration using CuSO₄ Magnesium-Copper Exchange Titration

Victor vmr Rodriguez¹

¹Independent Researcher

Victor vmr Rodriguez

Independent Researcher

ABSTRACT

This protocol is designed to be able to extract and analyze the concentration of chlorophyll within a sample on a molecular level. The procedures of this protocol require using copper II sulfate and hydrochloric acid as a means of titration to determine the approximate level a chlorophyll within a given sample.

GUIDELINES

For proper extraction in titration of Chlorophyll concentration you must be able to measure and transfer liquids within a hundred micro-liters ensure that samples are separated and free of contaminants.

MATERIALS

MATERIALS

☒ Magnesium sulfate heptahydrate **Contributed by users**

☒ Hydrochloric Acid **Contributed by users**

☒ Copper (II) sulfate pentahydrate **Bio Basic Inc. Catalog #CDB0063.SIZE.500g**

☒ Acetone **Nacalai Tesque Inc. Catalog #00310-95**

☒ Distilled Water **Thermo Fisher Catalog #15230196**

STEP MATERIALS

☒ Acetone **Nacalai Tesque Inc. Catalog #00310-95**

☒ Magnesium sulfate heptahydrate **Contributed by users**

☒ Magnesium sulfate heptahydrate **Contributed by users**

☒ Acetone **Nacalai Tesque Inc. Catalog #00310-95**

☒ Hydrochloric Acid **Contributed by users**

☒ Copper (II) sulfate pentahydrate **Merck MilliporeSigma (Sigma-Aldrich) Catalog #209198**

☒ Acetone **Nacalai Tesque Inc. Catalog #00310-95**

OPEN ACCESS

DOI:

dx.doi.org/10.17504/protocols.io.7xrhpm6

Protocol Citation: Victor vmr Rodriguez 2019. Determining Chlorophyll Concentration using CuSO₄ Magnesium-Copper Exchange Titration.

protocols.io

<https://dx.doi.org/10.17504/protocols.io.7xrhpm6>

License: This is an open access protocol distributed under the terms of the [Creative Commons Attribution License](#), which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited

Protocol status: In development

We are still developing and optimizing this protocol

Created: Oct 03, 2019

Last Modified: Oct 04, 2019

PROTOCOL integer ID: 28369

Keywords: chlorophyll, botany, plant biology, copper II sulfate

SAFETY WARNINGS

- ! This protocol requires the use of strong corrosive acids, corrosive and flammable solvents, and require the extraction of pigments that may stain clothing. Proper lab coat, eye protection, gloves in ventilation are required to conduct is chlorophyll extraction in titration protocol. Also, care must be taken to ensure that all materials used are disposed properly, as many of the chemicals may be hazardous to health and environment.

BEFORE START INSTRUCTIONS

In order to perform this chlorophyll extraction protocol you will need the following materials and chemicals:

the materials listed are based on one single sample, in must be multiplied based on the number of samples you would like to test

Two 20 ml (minimum) test tube

Ten 5 ml glass vials

A minimum of ten 200 micro-liters PCR tubes (although more may be needed based on the accuracy necessary for the procedures performed)

A 100 to 1000 micro-liter adjustable pipette

30 disposable 1000 micro-liter pipette tips (Number varies based on need and mistakes)

1200 g Centrifuge

One 20 micron filtration filter paper

Mortar and Pestle

Transfer Pipettes (as needed for contamination prevention)

10 ml Graduated Cylinder

Test tube stirrers

Approximate protocol time: 6 hours total - 3 hours Preparation - 3 Hours Sample Sit time

100ul per PCR Tube Copper II sulfate Solution 60mg/ml Distilled H2O

As needed Distilled water

9 ml Acetone

1g Magnesium sulfate

2g of sample

Preparation of Sample and Extraction

1 Add 2 g of sample **plant matter** or other test article to a mortar

1m

- 2 Add 1 g of Magnesium sulfate heptahydrate **Contributed by users** crystals to the mortar. 1m
- 3 Grind the sample item and Magnesium sulfate heptahydrate **Contributed by users** together until the leaves are ground up, the magnesium sulfate is completely dissolved into the plant liquid and a liquid starts to form. 5m
- 4 Transfer as much of the solid and liquid slurry to a 20ml test tube. 1m
- 5 Add 1 mL of Acetone **Nacalai Tesque Inc. Catalog #00310-95** to the mortar to wash up and collect as much of the sample as possible into the 20ml test tube. 2m
- 6 Add 8 mL Acetone **Nacalai Tesque Inc. Catalog #00310-95** to the 20ml test tube. This is the **raw mix**. 1m
- 7 Use a glass stirrer to stir the **raw mix**. Take care to press the plant slurry down to the base of the test tube, then agitate the slurry again to mix. Repeat this many times for 00:02:00 minutes. Ensure that the acetone is fully incorporated into the slurry. 2m
- 8 Let the **raw mix** stand for 00:05:00 minutes. 5m
- 9 Use a glass stirrer to stir the **raw mix** again for 00:01:00 minute. 1m

- 10 Using a **20 micron retention filtration paper** to filter the raw mix into another test tube. Ensure to use a new filter paper for each separate sample. This is the **refined mix**. 2m

Dilution Phase

- 11 Add 1 mL Hydrochloric Acid **Contributed by users** 1m
- to the **refined mix** and mix with clean glass stirrer for 00:01:00 minute to create the **Acid Activated Mix**.
- 12 Let the **Acid Activated Mix** stand for 00:02:00 minutes. 2m
- 13 Determine the resolution needed for the concentration. There resolution is the interval of dilution. For example, a 10% resolution will result in a dilution interval of 100%, 90%, 80% etc. The smaller the resolution the narrower the margin of error is. For this protocol, the resolution used will be 10%. 1m
- 14 Label ten 5ml vials in descending order from 100% to 10%. 1m
- 15 Add **Acid Activated mix** to each 5ml vial at volume based on the formula $V_{ol} = 1000ul * P_{er}$ with P_{er} representing the percentage written on each vial. Use the percentage or decimal, not whole number. **Ex. 10% = .10** 5m
- 16 Add Acetone **Nacalai Tesque Inc. Catalog #00310-95** 5m
- to each 5ml vial at volume based on the formula $V_{ol} = 1000ul * (1 - P_{er})$ with P_{er} representing

the percentage written on each vial. Use the percentage or decimal, not whole number. **Ex. 10%**
= .10

- 17 Gently stir each vial by rotating them in a circular motion for 00:00:30 seconds each. 5m
- 18 Label The proper amount of PCR tubes corresponding to the percentages listed on the vials. 1m
- 19 Using an adjustable pipette, dispense 100 μL of **Acid Activated Mix** from each vial to the corresponding PCR Tube. Ensure to change the pipette tip for every transfer in order to ensure contamination prevention. 5m

Copper Transfer Phase

- 20 Prepare a **60mg CuSO_4 /ml H_2O** concentration solution of Copper (II) sulfate pentahydrate **Merck MilliporeSigma (Sigma-Aldrich) Catalog #209198** to be referred to as the **copper (II) sulfate Solution**. 5m
- 21 Add 100 μL **copper (II) sulfate Solution** to each PCR tube. 5m

Test Phase

- 22 Close all PCR tubes and Centrifuge all tubes at 1200 x g for 2 minutes. 2m
- 23 In a PCR rack, let the samples stand for 6 hours 6h

24 Centrifuge all tubes at 1200 x g for 2 minutes.

2m

Results Phase

25 Observe which PCR is the lowest percentage concentration not to have any trace of precipitated **Copper (II) Sulfate**. (A blue crystalline precipitate at the bottom. Take note of the percentage number to be used as the variable **P₋**.)

1m

26 Assign the highest percentage value vial to have visible **copper (II) Sulfate** precipitate the variable **P₊**.

1m

27 Use this formula to calculate the mean value for the concentration in g/ml.

10m

$$\frac{(6 \times 10^{-4} \times 893.51)}{(P_{-} \times 159.609)} + \left(\frac{(6 \times 10^{-4} \times 893.51)}{(P_{+} \times 159.609)} - \frac{(6 \times 10^{-4} \times 893.51)}{(P_{-} \times 159.609)} \right) / 2 = \text{Con}$$

28 Use this formula to calculate the margin of error of the concentration.

10m

$$\left(\frac{(6 \times 10^{-4} \times 893.51)}{(P_{+} \times 159.609)} - \frac{(6 \times 10^{-4} \times 893.51)}{(P_{-} \times 159.609)} \right) / 2 = \text{Moe}$$

29 Write the final results as **Con ± Moe**

1m