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Case - Heavy Water Assays by GC-mass spectrometry

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We use this protocol and it's working

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Abstract

Summary:

Heavy water can be used as a tracer for estimating metabolic rates (*in vivo*, *in vitro*) such as total energy expenditure (TEE) and fractional synthesis rates (FSR; eg. protein, lipids, triglycerides). When using both $^2\text{H}_2\text{O}$ and H_2^{18}O (DLW), TEE is estimated from the total production of CO_2 as measured by the differences in decay rates of labeled the ^{18}O and ^2H in body water over time following a single bolus of DLW (1). $^2\text{H}_2\text{O}$ can be used to estimate fractional synthesis rates of metabolic reactions such as those associated with proteins, lipids, triglycerides, and cholesterol (2,4).

References:

1. Gas chromatography-mass spectrometry assay of the (18) O enrichment of water as trimethyl phosphate. Brunengraber DZ, McCabe BJ, Katanik J, and Previs SF. Anal Biochem 306: 278– 282 (2002).
2. Increased plasma membrane cholesterol in cystic fibrosis cells correlates with CFTR genotype and depends on de novo cholesterol synthesis. Fang D, West RH, Manson ME, Ruddy J, Jiang D, Previs SF, Sonawane ND, Burgess JD, Kelley TJ. Respir Res.; 11:61 (2010).
3. Triglyceride synthesis in epididymal adipose tissue: contribution of glucose and non-glucose carbon sources. Bederman IR, Foy S, Chandramouli V, Alexander JC, Previs SF. J Biol Chem.; 284(10):6101-8 (2009).
4. Novel application of the "doubly labeled" water method: measuring CO_2 production and the tissue-specific dynamics of lipid and protein in vivo. Bederman IR, Dufner DA, Alexander JC, Previs SF. Am J Physiol Endocrinol Metab.; 290(5):E1048-56 (2006).

Materials

MATERIALS

☒ Deuterium oxide 2H₂O (99 atom % excess); Merck MilliporeSigma (Sigma-Aldrich)

☒ H₂ ¹⁸O (95% atom excess) Isotec

Reagents and Materials:

Reagent/Material	Quantity Required	Vendor
Deuterium oxide, ² H ₂ O (99 atom % excess);		Sigma-Aldrich
NaOH	2μL	stock
H ₂ ¹⁸ O (95% atom excess)		Isotec (Miamisburg, OH)
acetone/acetonitrile solution	4μL	stock
Chloroform	200- 500 μL	stock
Phosphorus Pentachlorate; TMP*	3 mg	stock
Diethyl ether	120μl	stock

Analysis of ²H₂O and H₂ ¹⁸O: ²H labeling of body water is determined by exchange with acetone and the ¹⁸O labeling of body water is determined by conversion to trimethyl phosphate *(TMP; generated by reacting phosphoric acid with diazomethane).

Note:

Sigma-Aldrich,RRID:SCR_008988

Troubleshooting

- 1
1. Quantitation by $^2\text{H}_2\text{O}$ Standard Curve:
2. Standards are made from deuterium oxide (Aldrich 617385)
3. Pipette 10 μl of plasma or standard into Eppendorf
4. $^2\text{H}_2\text{O}$ standards
 - a. Blank (MQ H_2O)
 - b. 0.1% D_2O in (MQ H_2O)
 - c. 0.5% D_2O in (MQ H_2O)
 - d. 0.10% D_2O in (MQ H_2O)
 - e. 1.5% D_2O in (MQ H_2O)
 - f. 2.0% D_2O in (MQ H_2O)
 - g. 2.5% D_2O in (MQ H_2O)
 - h. 3.0% D_2O in (MQ H_2O)
 - i. 3.5% D_2O in (MQ H_2O)
 - j. 4.0% D_2O in (MQ H_2O)
5. Add 2 μl of a 10N NaOH solution to each sample or standard
6. Add 4 μl of acetone/acetonitrile solution (10 μl of acetone +200 μl of acetonitrile) to each sample
7. Be careful when taking samples out of the centrifuge to make sure that all of the drops are at the bottom of the tube
8. Cap and briefly centrifuge samples (~5sec) to ensure NaOH and acetonitrile react with sample
9. Let samples sit overnight (at least 10hrs)
10. Add 500 μl of chloroform to samples
11. Add ~ 50mg Na_2SO_4 salt to each sample
12. Centrifuge sample for 2 minutes
13. Pipette 100 μl of chloroform layer into glass insert, place inserts in GC vials and cap, then assay on a GC-MS system, EI mode (see below for parameters and references).

2 For doubly labelled water studies, such as total energy expenditure; TEE (1,4):

1. H₂¹⁸O standards:

- a. Blank (MQ H₂O)
- b. 0.01% H₂¹⁸O in (MQ H₂O)
- c. 0.05% H₂¹⁸O in (MQ H₂O)
- d. 0.10% H₂¹⁸O in (MQ H₂O)
- e. 0.15% H₂¹⁸O in (MQ H₂O)
- f. 0.20% H₂¹⁸O in (MQ H₂O)
- g. 0.25% H₂¹⁸O in (MQ H₂O)
- h. 0.30% H₂¹⁸O in (MQ H₂O)
- i. 0.35% H₂¹⁸O in (MQ H₂O)
- j. 0.40% H₂¹⁸O in (MQ H₂O)

2. H₂¹⁸O assay: 5 ul of blood or plasma sample or standard into a 12 × 75-mm glass tube

3. Then add ~3 mg of PCI₅ to samples or standards to generate phosphoric acid; let stand for 20 min

4. To generate TMP, then react samples or standards by adding 300 µl of freshly prepared ethereal-diazomethane (to derivatize samples) and allow to stand at room temperature during reaction, until the ether evaporated; may use an additional 120µl of diethyl ether in hexane solution

5. TMP is extracted by addition of 150 ul of water and 600 µl of chloroform (1:4 ratio) followed by addition of 0.5 g Na₂ SO₄

6. Samples are then vigorously mixed, and a small aliquot of the chloroform is transferred to a GC-MS vial and assayed on GC-MS system, EI mode

Gas Chromatography Mass Spectrometry, GC-MS (EI mode): Acetone and the TMP derivatives are analyzed using an Agilent 5973N-MSD equipped with an Agilent 6890 GC system, and a DB-17MS capillary column (30 m x 0.25 mm x 0.25 µm). The mass spectrometer is operated in the electron impact mode (EI; 70 eV), (1,4).

7. Selective ion monitoring of mass-to-charge ratios (m/z)



- a. For ^2H enrichments monitor acetone (58, 59) and for ^{18}O enrichments monitor TMP (140,142)
- b. The ^2H enrichments are calculated from the $^2\text{H}_2\text{O}$ standard curve and the ^{18}O enrichments are calculated from the signal ratio $(142)/(142 + 140)$.