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Benzoyl-Hexose derivatization

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We use this protocol and it's working

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Abstract

Hexose derivatization is a crucial analytical procedure used to enhance the detection and quantification of hexose sugars in various samples by liquid chromatography-mass spectrometry (LC-MS).

Materials

- Ethyl acetate
- Water
- 1 M Potassium phosphate dibasic (K_2HPO_4)
- 8 M Sodium hydroxide (NaOH)
- Benzoyl chloride
- 1.4 M Phosphoric acid (H_3PO_4)
- 1 M Sodium hydroxide (NaOH)
- Acetonitrile
- 100 mM Ammonium formate (NH_4HCO_2)
- Vortex
- Centrifuge (capable of 16,000 g)
- Speed vacuum
- Mass spectrometer (for m/z detection)



- 1 Add  200 μL of a **1:1 ethyl acetate:water** to cells or beads after organelle IP such as LysolP
- 1.1 Vortex the mixture for  00:05:00 5m
- 1.2 Centrifuge at  16000 x g, 00:05:00 5m
- 2 Carefully remove the **upper layer** (ethyl acetate) and discard.
- 3 Transfer **the bottom (water) layer** to a new tube.
- 4 **Benzoyl Derivatization**
- 4.1 Add  20 μL of **1 M potassium phosphate dibasic (K_2HPO_4)**
- 4.2 Add  20 μL of **8 M NaOH**
- 4.3 Add  10 μL of **benzoyl chloride**
- 4.4 Vortex the mixture for  00:05:00 and then briefly spin down to collect the liquid. 5m
- 5 Add  10 μL of **1.4 M phosphoric acid (H_3PO_4)**
- 6 Add  500 μL of **ethyl acetate**
- 7 Vortex for  00:03:00 and Centrifuge at  16000 x g, 00:05:00 8m



- 8 Transfer  400 μL of the **upper layer** (ethyl acetate) to a new tube.
- 9 Add **100 μL of 1 M NaOH** to the ethyl acetate extract.
- 10 Vortex for  00:02:00 and then centrifuge at  16000 x g, 00:05:00 7m
- 11 Transfer  100 μL of the **upper layer** (ethyl acetate) to a new tube.
- 12 **Repeat the process:**
Add  100 μL of **1 M NaOH** to the remaining ethyl acetate solution.
- 13 Vortex for  00:02:00 Centrifuge at  16000 x g, 00:05:00 7m
- 14 Transfer  100 μL of **upper layer** to the same tube as the previous  100 μL , for a total of  200 μL of ethyl acetate.
- 15 **Drying and Reconstitution**
 - 15.1 Dry the 200 μL of ethyl acetate in a **speed vacuum**.
 - 15.2 Reconstitute the dried sample in  50 μL of a **9:1 acetonitrile:water containing 100 mM ammonium formate (NH_4HCO_2)**.
- 16 **Mass Spectrometry Monitoring**
 - Analyze the sample for **parent 718 m/z $[\text{M}+\text{NH}_4]^+$** and the following product ions:
 - **231**
 - **105**
 - **579.1**